

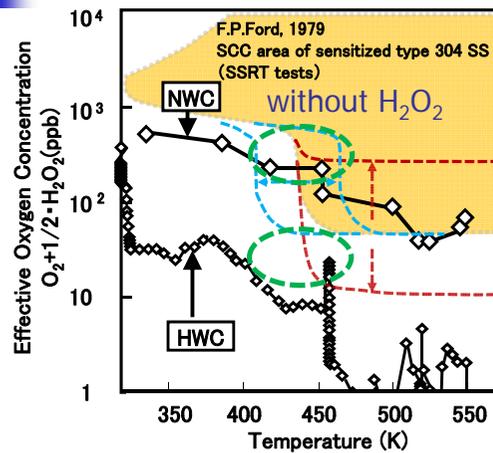
423および453Kの高温純水中における鋭敏化ステンレス鋼のSCC感受性に及ぼす過酸化水素の影響

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Introduction

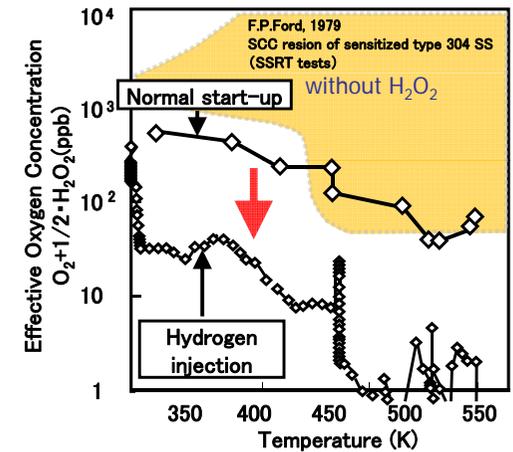


H₂O₂ changes border ?

SCC susceptibility at 423K and 453K with H₂O₂ ?

Introduction

- At plant start-up in BWR hydrogen peroxide and oxide concentration increase and dynamic distortions occur
- ↓
- Severe condition for structural materials
- ↓
- Hydrogen injection during start-up



Purpose and Experiments

Purpose

- Effect of hydrogen peroxide on SCC susceptibility
- Effect of hydrogen peroxide on oxide film

Experiments

- SSRT tests of sensitized type 304 SS
- TEM observation, Laser Raman spectrometry of oxide film

Experiment

Material

Type	C	Si	Mn	P	S	Ni	Cr	Fe
304 SS	0.050	0.62	1.00	0.027	0.005	8.16	18.25	Bal.

Sensitization; 893 K × 24hr, air cooled

Surface; #600 polished

Tests

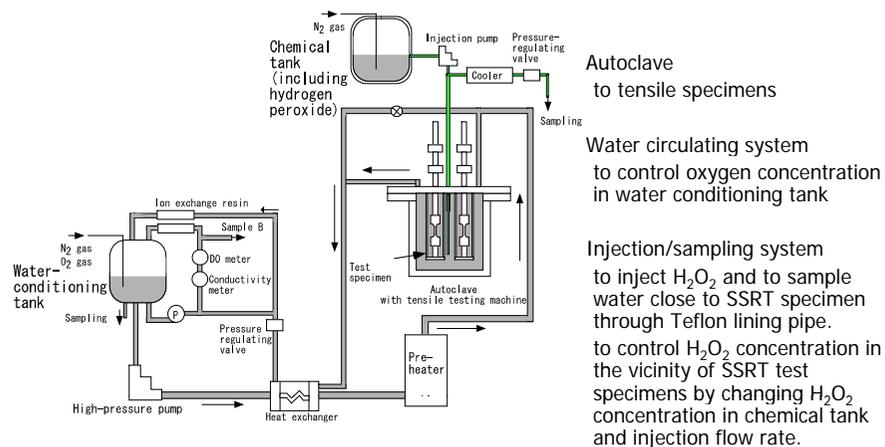
SCC tests	SSRT (plate; gauge 20mm(l) × 5mm(w) × 2mm(t))
Strain rate	4.2×10^{-7} /sec

Experiment

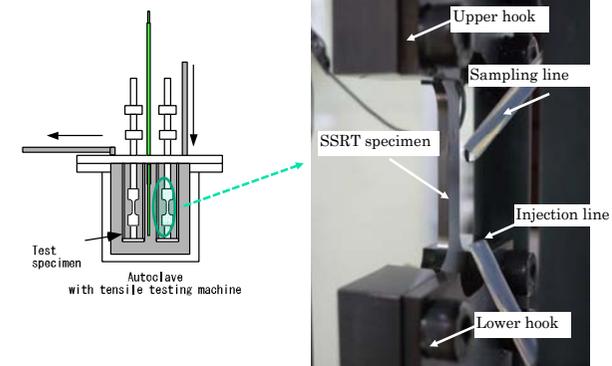
Water condition

Temp. (K)	EOC (ppb)	DO, at 293 K (ppb)	H ₂ O ₂ at 293 K (ppb)	Number of specimens
453	400	400	0	4
	400	25	750	4
	45	45	0	4
	45	5	80	4
	27	10	33	4
423	20	8	24	4
	400	50	700	4
	47	10	74	4

Test loop

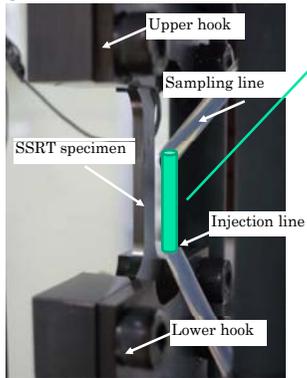


Overview of specimen setting



H₂O₂ was injected from the bottom of SSRT specimen
 H₂O₂ was sampled from the middle of SSRT specimen

Decomposition in sampling and injection line



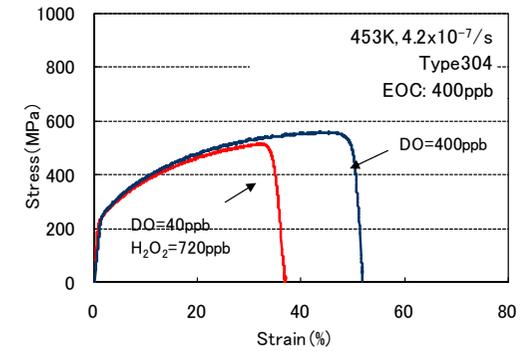
Directly connecting injection line to sampling line, 99.5% of H₂O₂ concentration in chemical tank was detected in sampled water at 453 K.

Decomposition of H₂O₂ in injection and sampling line was very low.

During SSRT testing, 85% of H₂O₂ in sampling line at 293 K were detected in sampled water at 453 K.

720~765 ppb H₂O₂ at 293 K
 ⇒ 615~657 ppb H₂O₂ at 453 K

Results of SSRT tests, EOC=400ppb



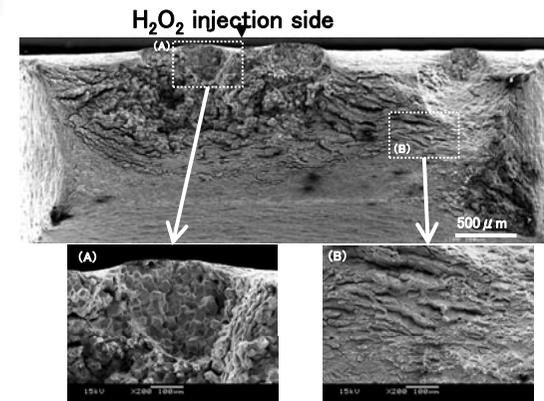
Specimens tested under H₂O₂ injection condition were decreased in elongation.

Tensile properties

EOC	H ₂ O ₂ (453 K)	H ₂ O ₂ (453 K)	0.2 % proof stress	Tensile strength	Elongation	Reduction of area
(ppb)	(ppb)	(ppb)	(MPa)	(MPa)	(%)	(%)
400	0	0	248	580	50	63
			231	546	55	68
			237	560	52	71
			228	571	53	70
400	720	617	227	516	37	51
	764	657	228	395	16	26
	736	563	225	412	17	15
	748	642	226	466	27	48

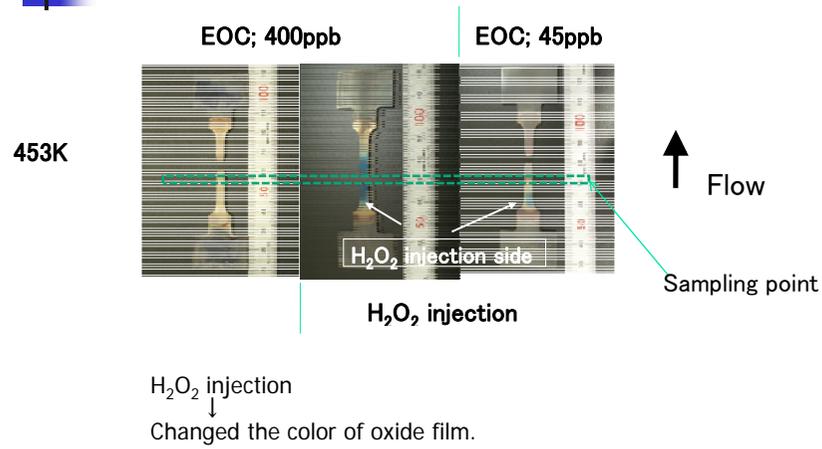
H₂O₂ injection reduced TS, EL and R.A.

Appearance of fracture surface after SSRT test.

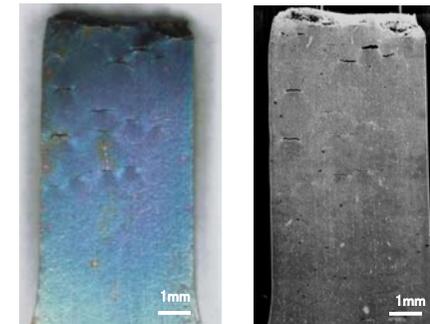


453K, EOC; 400ppb, H₂O₂ injection
 IGSCC fractures were observed.

Appearance of specimens after SSRT tests



Appearance of specimens after SSRT tests



Many SCC cracks were observed on H₂O₂ injection side.
And counted number of SCC crack

Rate of SCC on fracture surface Number of cracks on the surface

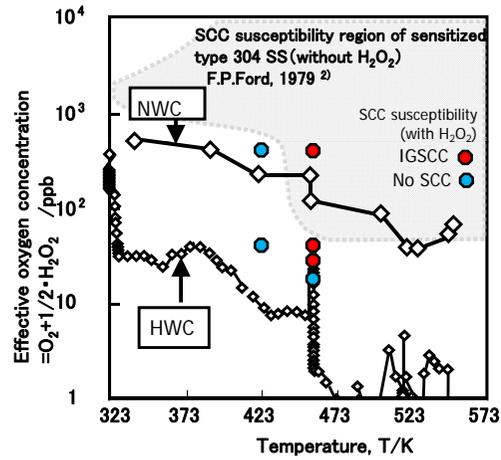
EOC (ppb)	H ₂ O ₂ (RT) (ppb)	H ₂ O ₂ (453K) (ppb)	SCC	Rate of SCC (%)	Number of SCC	
					H ₂ O ₂ side	Another side
400	0	0	No	0	0	0
			Observed	3.9	0	0
			No	0	0	0
			No	0	0	0
400	720	617	Observed	22.7	39	0
	764	657	Observed	63.7	32	0
	736	563	Observed	67.4	47	0
	748	642	Observed	41.1	6	0

SCC cracks were observed on only H₂O₂ injection side

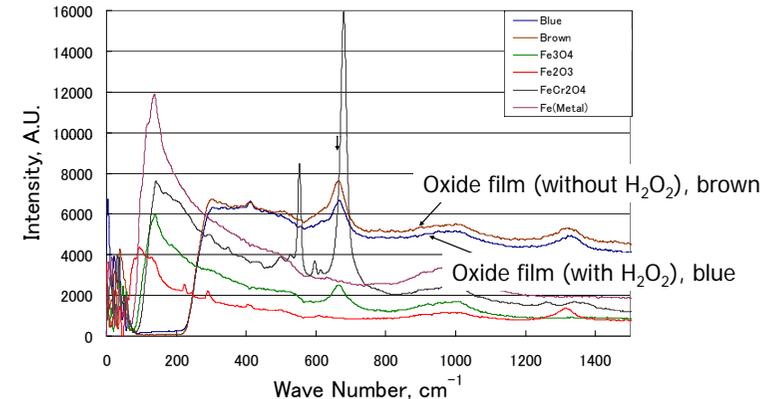
SSRT test results

Temperature (K)	EOC (ppb)	H ₂ O ₂ injection	SCC occurrence / specimens
			*/4
453	400	No	1/4
		Yes	4/4
	45	No	0/4
		Yes	1/4
423	27	Yes	2/4
		Yes	0/4
	400	Yes	0/4
		Yes	0/4

SSRT test results

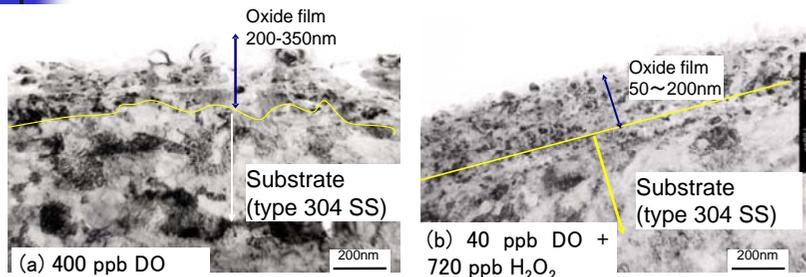


Laser Raman spectrometry



Oxide films are mainly consisted of Fe₃O₄ (Magnetite)
No difference between oxide films tested under water condition with or without H₂O₂

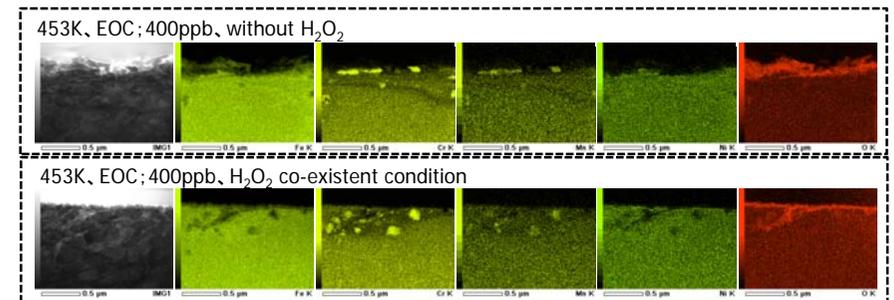
TEM observation



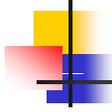
TEM micrographs of cross sections of sensitized type 304 SS tested under the conditions of 400 ppb EOC without H₂O₂ (a) and with H₂O₂ (b) at 453 K.

H₂O₂ injection makes planar surfaces of oxide film.

TEM observation

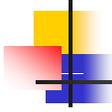


H₂O₂ injection thins oxide film thickness.



Conclusion

- The addition of H_2O_2 increases the SCC susceptibility of sensitized type 304 SS at 453 K.
- SCC susceptibility was not observed under 400 ppb EOC condition with and without H_2O_2 at 423 K.
- The addition of H_2O_2 changes the surface shape of oxide film in sensitized type 304 SS, from wavy to planar and thins oxide film thickness.



Thanks for your attention !